# **Technical Data Sheet**

# Chemical resistance of Grilon, Grilamid, Grilamid TR, Grivory GV and Grivory HTV

Plastics play a key role, both in industry and in everyday life. It is, however, extremely important that a specific plastic, unaffected by the environment surrounding it, is chosen for each application.

Generally speaking, polyamides are very resistant to all types of chemicals. Apart from concentrated acids, very few reagents attack polyamides.

The following table showing the chemical resistance of Grilamid, Grilamid TR, Grivory GV and Grilon offers guidance on the choice of polyamides for particular end uses.



The following table gives an indication of the chemical resistance of Grilon (polyamide 6 and 66), Grilamid (polyamide 12), Grilamid TR (transparent polyamide 12), Grivory GV (partially aromatic polyamide) and Grivory HTV (Polyphthalan 1). The chemical resistance was established by exposing test samples 1 mm thick, to each of the chemicals for a period of 12 months at room temperature. The results are valid both for unreinforced and for glass fibre reinforced products.

## Key:

- Resistant. Negligible, reversible or no changes in mass and dimensions. Example: Grilon unaffected by aqueous and alcoholic media.
- Limited resistance. Considerable dimensional changes, and possibly irreversible changes in properties after prolonged contact. Consultation advisable before use.
- Not resistant. May be used under certain conditions (brief contact).
- Soluble or attacked after brief contact.

Certain additives, particularly plasticizers, may be dissolved out into the medium. Absorption of the medium is generally sufficient, however, to compensate for any resultant loss in flexibility.

The data regarding chemical resistance refers to stress-free products. Stresses in parts of Grilamid TR can lead to cracking when coming into contact with specific solvents. Particular information can be found in the section «Environmental Stress Cracking».

## **Environmental Stress Cracking of Grilamid TR**

Amorphous thermoplastics such as Grilamid TR can develop stress cracking when exposed to certain media. Components are more likely to develop stress cracking symptoms when they are subjected to external stresses, or when, through unsuitable processing, they have high internal stresses.

Grilamid TR 55, Grilamid TR 70 LX and Grilamid TR 90 are not resistant to the following chemicals and stress cracking may occur: Benzyl alcohol, butanol, butylene glycol, ethanol, isopropanol, methanol, phenyl ethyl alcohol, propanol.

Grilamid TR 55 and Grilamid TR 70 LZ have limited (short term) chemical resistance to the following chemicals but stress cracking may occur under conditions of high internal or external stress: Acetone, amyl acetate, benzaldehyde, butyl acetate, cyclohexanone, diethyl ether, etheric oils, ethyl acetate, isopropanol 80%, methyl ethyl ketone, phenyl ethyl alcohol, pyridine, tetrahydrofuran.

Grilamid TR 90 has limited (short term) resistance to the following chemicals, but stress cracking may occur in: amyl acetate, benzaldehyde, butyl acetate, cyclohexanone, etheric oils, phenyl ethyl alcohol, pyridine.

The chemical resistance is dependent both on the temperature and the stress condition of the finished component. The suitability of any material for a specific application must be confirmed by a practical test.

Medium	Chemical formula	Concentration	Resistar				
			Grilon	Grilamid	Grilamia IK	Grivory GV	Grivory HTV
Acetaldehyde	CH <sub>3</sub> -CHO	40 % ag. soln.	••	•••	••	••	••
Acetamide	$CH_3-CO-NH_2$	50 % aq. soln.	•••	•••	•••	$\bullet \bullet \bullet$	•••
Acetic acid	CH <sub>3</sub> COOH	10 % aq. soln.		••	$\bullet \bullet$	•	••
Acetic acid	CH <sub>3</sub> COOH	40 % aq. soln.	$\circ$	•	•	0	•
Acetic acid	CH <sub>3</sub> COOH	technically pure	$\circ$	•	0	0	•
Acetic anhydride	CH <sub>3</sub> -CO-O-OC-CH <sub>3</sub>	technically pure	$\circ$	••	•	•	0
Acetone	CH <sub>3</sub> -CO-CH <sub>3</sub>	technically pure	•••	•••	•	••	$\bullet \bullet \bullet$
Allyl alcohol	H <sub>2</sub> C=CH-CH <sub>2</sub> -OH	technically pure	$\bullet$	•	0	••	••
Aluminium salts	_	*, aq. soln.	•••	•••	•••	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Alums	$K_2SO_4 - Al_2(SO_4)_3 \cdot 12 H_2O$	*, aq. soln.	$\bullet$	•••	•••	••	$\bullet \bullet \bullet$
Ammonia	NH <sub>3</sub>	10 % ag. soln.	•••	•••	•••	•••	$\bullet \bullet \bullet$
Ammonia	NH <sub>3</sub>	*, gaseous	•••	•••	•••	•••	$\bullet \bullet \bullet$
Ammonium chloride	NH <sub>4</sub> Cl	10 % aq. soln.	•••	•••	•••	•••	•••
Ammonium salts		*, technically pure	$\bullet \bullet$	•••	••	••	•••
Amyl acetate	$CH_3(CH_2)_4$ $-OOCCH_3$	technically pure	•••	••	•••	•••	•••
Amyl alcohol	$CH_3(CH_2)_3 - CH_2 - OH$	technically pure	•••	•••	0	•••	•••
Aniline	$C_6H_5-NH_2$	technically pure	••	••	0	••	••

<sup>\*</sup> signifies data valid for all concentrations

Medium	Chemical formula	Concentration	Grilon	Grilamid		esistance ilamid TR Grivory GV Grive		
Anisole	C <sub>6</sub> H <sub>5</sub> -O-CH <sub>3</sub>	technically pure	•••	•••	•••	•••	•••	
Aqua regia	HNO <sub>3</sub> + HCI	technically pure	$\circ$	$\circ$	$\circ$	$\circ$	$\circ$	
Aspirin	_	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Attar of roses (Rose oil)	_	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Barium salts	_	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Battery acid	$H_2SO4$	36 % aq. soln.	•	$\bullet$	$\bullet$	•		
Beer	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Benzaldehyde	C <sub>6</sub> H <sub>5</sub> CHO	technically pure	•	$\bullet$		•		
Benzoic acid	$C_6H_5$ —COOH	*, aq. soln.	$\bullet$	$\bullet$	$\bullet$	$\bullet$	$\bullet$	
Benzene	$C_6H_6$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Benzyl alcohol	$C_6H_5-CH_2OH$	technically pure	•	•	•	•		
Bitumen	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Borax	$Na_2B_4O_7$	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Boric acid	$H_3BO_3$	10 % aq. soln.	$\bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet$	$\bullet$	
Brake fluid (DOT 4)	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Brandy	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Bromine	Br <sub>2</sub>	*	•	•	0	•		
Butane	$C_4H_{10}$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Butanol	$C_4H_9OH$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	0	$\bullet$	$\bullet \bullet \bullet$	
Butter	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Butter milk	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Butyl acetate	CH <sub>3</sub> COOCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Butyric acid	CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> —COOH	technically pure	$\bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet$	$\bullet$	
Butylene glycol	HO-CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> -OH	technically pure	$\bullet$	$\bullet \bullet \bullet$	0	$\bullet$	$\bullet$	
Calcium chloride	CaCl <sub>2</sub>	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Calcium chloride	CaCl <sub>2</sub>	20 % alcoholic soln.	$\circ$	•	$\circ$	•		
Camphor	_	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Carbon disulphide	$CS_2$	100 %	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Carbon tetrachloride	CCI <sub>4</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Caustic soda	NaOH	40 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Chlorinated lime	$Ca(CIO)_2$	*, aq. soln.	$\circ$	0	0	$\circ$	$\circ$	
Chlorine	$Cl_2$	technically pure	$\circ$	0	0	$\circ$	$\circ$	
Chlorine gas	$Cl_2$	< 5 %, gaseous		$\bullet$	•	$\bullet$	$\bullet$	
Chlorine water	_	< 5 %, aq. soln.		$\bullet$	•	$\bullet$	$\bullet$	
Chloroacetic acid	CICH <sub>2</sub> COOH	10 %, technically pure	$\circ$	$\circ$	$\circ$	$\circ$	0	
Chlorobenzene	$C_6H_5-CI$	technically pure	$\bullet \bullet \bullet$	•	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Chlorobrommethane	CH <sub>2</sub> ClBr	technically pure	$\bullet$	$\bullet$	•	$\bullet$	$\bullet$	
Chloroform	CHCl <sub>3</sub>	technically pure		•	•	•		
Chromic acid	$H_2CrO_4$	10 % aq. soln.	$\circ$	•	•	$\circ$	$\circ$	
Chromic acid	$H_2CrO_4$	1 % aq. soln.		$\bullet$	•	•		
Chromic/sulphuric acid	$H_2SO_4/CrO_3$	*, aq. soln.	$\circ$	0	0	$\circ$	$\circ$	
Chromium salts	_	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Coca-Cola	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Cocoa	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Coffee	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Copper salts	_	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	
Cresol	$H_3C-C_6H_4-OH$	technically pure	0	0	0	0	0	

Medium	Chemical formula	Concentration	Grilon	Resistance on Grilamid Grilamid TR Grivory GV G			
Cyclohexane	C <sub>6</sub> H <sub>12</sub>	technically pure	•••	•••	•••	•••	•••
Cyclohexanol	C <sub>6</sub> H <sub>11</sub> OH	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Cyclohexanone	C <sub>6</sub> H <sub>10</sub> O	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Decalin	C <sub>10</sub> H <sub>18</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Dibutyl phthalate	$C_6H_4 - (COOC_4H_9)_2$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Diesel	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Diesel oil	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Diethyl ether	$CH_3-CH_2-O-CH_2-CH_3$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Dimethyl formamide	$HCON-(CH_3)_2$	technically pure	$\bullet \bullet \bullet$	$\bullet$	$\circ$	$\bullet$	$\bullet \bullet \bullet$
Dioctyl phthalate	$C_6H_4-(COOC_8H_{17})_2$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Dioxane	$C_4H_8O_2$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Edible fats and oils	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Ethanol	CH <sub>3</sub> CH <sub>2</sub> OH	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\circ$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Ether	CH <sub>3</sub> CH <sub>2</sub> -O-CH <sub>2</sub> CH <sub>3</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Ethyl acetate	CH <sub>3</sub> COOCH <sub>2</sub> CH <sub>3</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Ethylene chloride	CICH <sub>2</sub> -CH <sub>2</sub> CI	technically pure	$\bullet \bullet \bullet$	$\bullet$	•	$\bullet$	$\bullet \bullet \bullet$
FAM B	_	technically pure	$\bullet \bullet \bullet$	$\bullet$	$\circ$	$\bullet$	$\bullet \bullet \bullet$
Formaldehyde (Formalin)	HCHO	40 % aq. soln.	•	$\bullet$	$\bullet$	•	$\bullet \bullet$
Formamide	HCONH <sub>2</sub>	technically pure	$\bullet$	$\bullet$	$\bullet$	••	$\bullet \bullet$
Formic acid	HCOOH	10 % aq. soln.	•	•	$\bullet$	•	$\bullet \bullet$
Formic acid	HCOOH	40 % aq. soln.	$\circ$	•	•	•	•
Formic acid	HCOOH	85 % aq. soln.	$\circ$	•	$\circ$	0	0
Freon	partially halogenized	commercial grade	•	•	•	•	•
	fully halogenized	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Freon 12	$CF_2CI_2$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Freon 22	CHF <sub>2</sub> Cl	technically pure	•	•	•	•	•
Fruit juices	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Fuel C	free from lead	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Fuel oil	_	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Furfurol	$C_4H_3O$ —CHO	technically pure	$\bullet$	$\bullet$	$\bullet$	$\bullet$	$\bullet$
Glycerine	$C_3H_8O_3$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Glycol	HO-CH <sub>2</sub> CH <sub>2</sub> -OH	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Heptane	C <sub>7</sub> H <sub>16</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Hexane	C <sub>6</sub> H <sub>14</sub>	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Hydraulic fluid	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Hydrochloric acid	HCI	10 % aq. soln.	$\circ$	•	$\bullet$	$\circ$	$\circ$
Hydrochloric acid	HCI	1 % aq. soln.	•	$\bullet$	$\bullet \bullet \bullet$	•	•
Hydrogen fluoride	HF	40 % aq. soln.	$\circ$	$\circ$	$\circ$	$\bigcirc$	$\circ$
Hydrogen peroxide	$H_2O_2$	30 % aq. soln.	$\circ$	$\circ$	$\circ$	$\bigcirc$	$\circ$
Hydrogen peroxide	$H_2O_2$	10 % aq. soln.	•	$\bullet$	$\bullet$	•	•
Hydrogen peroxide	$H_2O_2$	2 % aq. soln.	•	$\bullet$	$\bullet$	•	$\bullet$
Hydrogen sulphide	$H_2S$	< 5 %, gaseous	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Ink	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
lodine tincture	$J_2$	*, alcoholic soln.	$\circ$	$\circ$	$\circ$	$\circ$	$\circ$
Iron salts	_	20 % aq. soln. neut.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Iron salts	_	20 % aq. soln. acid.	$\circ$	•	•	•	•
Isooctane	$(CH_3)_3C-CH_2CH(CH_3)_2$	technically pure					

Medium	Chemical formula	Concentration	Grilon	Grilamid	<b>Resistar</b> Grilamid TR		Grivory HTV
leannand glackel		tachnically pure	•			,	,
Isopropyl alcohol Lactic acid	(CH <sub>3</sub> ) <sub>3</sub> -CHOH CH <sub>3</sub> CH(OH)-COOH	technically pure 90 % aq. soln.	0		0	0	0
Lactic acid	CH <sub>3</sub> CH(OH)-COOH	50 % aq. soln.					
Lactic acid	CH <sub>3</sub> CH(OH)-COOH	5 % aq. soln.					
Lanolin		commercial grade					
Lead salts	_	technically pure					
Lemon juice	_	*, commercial grade					
Linseed oil	_	commercial grade					
Liqueurs	_	commercial grade					
Lubrications oils,		commercial grade					
greases, soaps	_	commercial grade					
Magnesium hydroxide	$Mg(OH)_2$	10 % aq. soln.					
Magnesium salts		10 % aq. soln.					
Mercury	Hg	technically pure					
Mercury salts		*, aq. soln., neutral					
Methanol	CH <sub>3</sub> OH	technically pure			0		•••
Methylene chloride	$CH_2CI_2$	technically pure					
Methylethyl ketone	$CH_3-CO-CH_2-CH_3$	technically pure					
Milk		commercial grade					
Mineral oils	_	commercial grade					
Motor fuels	_	commercial grade					
Naphthalene	C <sub>10</sub> H <sub>8</sub>	technically pure					
Nickel salts		*, aq. soln.					
Nitric acid	$HNO_3$	*, aq. soln.	0	0	•	0	0
Nitrobenzene	$C_6H_5NO_2$	technically pure		•	••		•
Nitromethane	$CH_3NO_2$	technically pure					
Octane	C <sub>8</sub> H <sub>18</sub>	technically pure					
Oil (No. 3 ASTM)	— —	commercial grade					
Oil of lavendar	_	commercial grade			•		
Oil of pine needle	_	technically pure			•••		
Oil of turpentine	_	technically pure					
Oleic acid	_	technically pure					
Oleum	$H_2SO_4 + SO_3$	technically pure	0	0	0	0	0
Olive oil	—	commercial grade					
Oxalic acid	HOOC-COOH	10 % aq. soln.	••	•••	•••	••	••
Ozone	$O_3$	*, gaseous	•	•	•	•	•
Ozone	$O_3$	< 1 ppm, gaseous					
Paraffin oil	<del>_</del>	technically pure					
Peanut oil	_	commercial grade			•••		
Peppermint oil	_	technically pure	••	••	••	••	••
Perchlorethylene	$Cl_2C=CCl_2$	technically pure					
Petrol (unleaded, Esso)	— · · · · · · · · · · · · · · · · · · ·	commercial grade					
Petroleum	_	technically pure					
Petroleum ether	_	technically pure					
Phenol	C <sub>6</sub> H <sub>5</sub> OH	*, aq. soln.	•	•	•	•	•
Phenylethyl alcohol	$H_3C - CH(C_6H_5) - OH$	technically pure		•	•	•	••
Phosphoric acid	$H_3PO_4$	50 % aq. soln.	•	•	•	•	•
ospone deld	4	0 0 70 44. 00111.	_	_	•	•	•

Medium	Chemical formula	Concentration	Grilon	Grilamid	<b>Resistan</b> Grilamid TR		Grivory HTV
Phosphoric acid	H <sub>3</sub> PO <sub>4</sub>	10 % aq. soln.	•	••	••	•	•
Plasticizers							
(phthalates, phosphates)	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potash	K <sub>2</sub> CO <sub>3</sub>	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potassium bromide	KBr	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potassium chlorate	KCIO <sub>3</sub>	7 % aq. soln.		$\bullet$	$\bullet \bullet$	•	•
Potassium hydroxide	KOH	50 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potassium iodide	KJ	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potassium nitrate	$KNO_3$	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Potassium permanganate	$KMnO_4$	1 % aq. soln.	$\circ$	$\circ$	$\circ$	$\circ$	$\circ$
Potassium sulphate	$K_2SO_4$	10 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Propane	$C_3H_8$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••
Propanol	$C_3H_7OH$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet$	$\circ$	$\bullet$	$\bullet \bullet \bullet$
Pyridine	$C_5H_5N$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Pyrocatechol	$HO-C_6H_4-OH$	6 % aq. soln.	•	••	0	•	•
Resorcinol	$HO-C_6H_4-OH$	technically pure	0	0	0	0	$\circ$
Resorcinol	$HO-C_6H_4-OH$	*, alcoholic soln.	0	0	0	0	0
Rum	_	commercial grade	•••	•••	••	•••	•••
Salicylic acid	$HO-C_6H_4-COOH$	technically pure	•••	•••	•••	•••	•••
Silicone oils	<u> </u>	technically pure			•••		
Silver salts	_	*, aq. soln.					
Soap solution	_	10 % aq. soln.					
Sodium bicarbonate	NaHCO <sub>3</sub>	*, aq. soln.					
Sodium bisulphite	NaHSO <sub>3</sub>	10 % aq. soln.					
Sodium bromide	NaBr	10 % aq. soln.					
Sodium carbonate	Na <sub>2</sub> CO <sub>3</sub>	10 % aq. soln.					
Sodium chloride	NaCl	*, aq. soln.					
Sodium chlorite	NaClO <sub>2</sub>	5 % aq. soln.					
Sodium hydroxide	NaOH	40 % aq. soln.					
Sodium hypochlorite	NaOCI	5 % aq. soln.					
Sodium nitrate	NaNO <sub>3</sub>	10 % aq. soln.					
Sodium nitrite	NaNO <sub>2</sub>	5 % aq. soln.					
Sodium perborate	- 1 Nul NO2	5 % aq. soln.					
Sodium phosphate	Na <sub>3</sub> PO <sub>4</sub>	10 % aq. soln.					
Sodium sulphate	$Na_2SO_4$	10 % aq. soln.					
Sodium sulphide	Na <sub>2</sub> S	10 % aq. soln.					
Sodium sulphite	Na <sub>2</sub> SO <sub>3</sub>	10 % aq. soln.					
•	- 0	· ·					
Sodium thiosulphite	$Na_2S_2O_3$	10 % aq. soln.					
Soya oil	_	commercial grade				•••	
Starch		*, aq. soln.				•••	
Styrene	$C_6H_5$ - $CH$ = $CH_2$	technically pure				•••	
Sugar	$C_6H_{12}O_6$	*, aq. soln.					
Sulphur	S	technically pure					
Sulphur dioxide	SO <sub>2</sub>	< 5 %	•	••	••	•	
Sulphuric acid	$H_2SO_4$	technically pure	0	•	•	0	0
Sulphuric acid	H <sub>2</sub> SO <sub>4</sub>	36 % aq. soln.	•			•	•
Sulphuric acid	$H_2SO_4$	10 % aq. soln.				•	•

Medium	Chemical formula	Concentration	0.1		Resistance Grilamid TR Grivory GV Grivory		
			Grilon	Grilamid	Grilamid IR	Grivory GV	Grivory HTV
Sulphuric acid	$H_2SO_4$	2 % aq. soln.	•	••	•••	•	•
Table salt	NaCl	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Tallow	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Tar	_	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Tartaric acid	HOOC-CH(OH)-CH(OH)-COOH	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Tea	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Tetrahydrofuran	$C_4H_8O$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Tetralin	$C_{10}H_{12}$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Thionyl chloride	SOCI <sub>2</sub>	technically pure	$\circ$	$\circ$	$\circ$	0	0
Toluene	$C_6H_5-CH_3$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Trichlorethylene	Cl <sub>2</sub> C=CHCl	technically pure	$\bullet \bullet$	$\bullet \bullet$	$\bullet \bullet$	$\bullet$	$\bullet$
Urea	$H_2N-CO-NH_2$	20 % aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Vaseline	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Vinegar	CH₃COOH	commercial grade	$\bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet$	$\bullet$
Water	H <sub>2</sub> O	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Water glass	_	*, aq. soln.	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Wax	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$
Wine	_	commercial grade	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Xylene	$H_3C-C_6H_4-CH_3$	technically pure	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	•••	$\bullet \bullet \bullet$
Zinc chloride	$ZnCl_2$	10 % aq. soln.	••	•••	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$	$\bullet \bullet \bullet$

The recommendations and data given are based on our experience to date. No liability can be assumed in connection with their usage and processing.

Domat/Ems, June 2001

# **EMS-GRIVORY** worldwide

www.emsgrivory.com

#### **Switzerland**

EMS-GRIVORY
Reichenauerstrasse
CH-7013 Domat/Ems
Tel. +41 81 632 78 88
Fax +41 81 632 74 01
a unit of EMS-CHEMIE AG
E-Mail: welcome@emsgrivory.com

## Germany

EMS-CHEMIE (Deutschland) GmbH Business Unit EMS-GRIVORY Warthweg 14 D-64823 Gross-Umstadt Tel. +49 6078 78 30 Fax +49 6078 783 416 E-Mail: welcome@de.emsgrivory.com

#### France

EMS-CHEMIE (France) S.A.
Division EMS-GRIVORY
73-77, rue de Sèvres
Boîte postale 52
F-92105 Boulogne-Billancourt
Tel. +33 1 41 10 06 10
Fax +33 1 48 25 56 07
E-Mail: welcome@fr.emsgrivory.com

#### **Great Britain**

EMS-CHEMIE (UK) Ltd.
Business Unit EMS-GRIVORY
Drummond Road
Astonfields Industrial Estate
GB-Stafford ST16 3HJ
Tel. +44 1785 607 580
Fax +44 1785 607 570
E-Mail: welcome@uk.emsgrivory.com

#### **United States**

EMS-CHEMIE (North America) Inc. Business Unit EMS-GRIVORY 2060 Corporate Way P.O. Box 1717 Sumter, SC 29151, USA Tel. +1 803 481 91 73 Fax +1 803 481 38 20 E-Mail: welcome@us.emsgrivory.com

# Taiwan

EMS-CHEMIE (Asia) Ltd.
Business Unit EMS-GRIVORY
36, Kwang Fu South Road
Hsin Chu Industrial Park
Fu Kou Hsiang, Hsin Chu Hsien
Taiwan, R.O.C.
Tel. +886 35 985 335
Fax +886 35 985 345
E-Mail: welcome@tw.emsgrivory.com

#### Japan

EC-SHOWA DENKO K.K.
Business Unit EMS-GRIVORY
Yutaka Bldg.
4-9-3 Taito
Taito-ku
110-0016, Tokyo, Japan
Tel. +81 3 3832 1501
Fax +81 3 3832 1503
E-Mail: welcome@jp.emsgrivory.com

